Microorganisms



1145. MS-MEDIUM (MODIFIED)

Anaerobic double distilled water	800.00	ml
MS Buffer (see below)	200.00	ml
Elemental sulphur	5.00	g
$NaS_2O_3 \times 5 H_2O$	2.00	g
$MgSO_4 \times 7 H_2O$	7.00	g
KCI	0.48	g
CaCl ₂ x 2 H ₂ O	0.40	g
MgCl ₂ x 6 H ₂ O	0.80	g
Solution A (see below)	20.00	ml
Solution B (see below)	1.50	ml
Solution D (see below)	10.00	ml

The medium is prepared anaerobically under nitrogen gas and dispensed into Hungate tubes or serum bottle sealed with rubber septa. The medium to gas phase ratio should be about 1:10 (v:v). Sterilise the medium by heating at 100° C for 3 hours on three consecutive days.

Final pH of the medium for DSM 19534 is pH 6.0-6.8

Final pH of the medium for DSM 19557 is pH 5.9-6.5

MS Buffer:

100 mM NaOH saturated with CO₂

Each of the solutions below were prepared with anaerobic double distilled water and prepared under constant gassing with nitrogen.

Solution A:

NH ₄ Cl	100.00	g
$MgCl_2 \times 6 H_2O$	100.00	g
CaCl ₂ x 2 H ₂ O	40.00	g
Distilled water	1000.00	ml
Adjust to pH 4 with HCl		

Solution B:

$K_2HPO_4 \times 3H_2O$	200.00	g
Distilled water	1000.00	ml

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Solution D - Trace mineral solution (Ferguson and Mah, 1983):

$\begin{array}{cccccccccccccccccccccccccccccccccccc$			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Na_2 -EDTA x 2 H_2O	500.00	mg
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Na_2WO_4 \times 2 H_2O$	30.00	mg
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	CoCl ₂ x 6 H ₂ O	150.00	mg
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	CuCl ₂ x 2 H ₂ O	20.00	mg
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	MnCl ₂ x 4 H ₂ O	100.00	mg
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	NiSO ₄ x 6 H ₂ O	20.00	mg
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	FeSO ₄ x 7 H ₂ O	100.00	mg
$\begin{array}{ccc} & & & & & & & & \\ H_3 B O_3 & & & & & & \\ Al Cl_3 \times 6 \ H_2 O & & & & & \\ Na Mo O_4 \times 2 \ H_2 O & & & & \\ \end{array} \qquad \qquad \begin{array}{ccc} & & & & & \\ 40.00 & mg \\ & & & & \\ 10.00 & mg \end{array}$	H_2SeO_3	10.00	mg
$AICI_3 \times 6 H_2O$ 40.00 mg $NaMoO_4 \times 2 H_2O$ 10.00 mg	ZnCl ₂	100.00	mg
$NaMoO_4 \times 2 H_2O$ 10.00 mg	H ₃ BO ₃	10.00	mg
	AICI ₃ x 6 H ₂ O	40.00	mg
Distilled water 1000.00 ml	NaMoO ₄ x 2 H ₂ O	10.00	mg
	Distilled water	1000.00	ml

Adjust pH to 3.0 with HCl